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From the information we just gathered about

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the H atom vs. the H ion, and the C atom vs. the C ion: An atom has a charge of zero. An ion is a particle with an overall positive or negative charge..

Compare the pictures of the atom and the $2+$ ion. If an ion has a $2+$ charge, what has been changed in the atom?

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Identify each of the

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following as an example of a Physical Property or a Chemical Property.

Chemistry States Of Matter Packet Answers Key

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This packet is designed to help you understand several concepts about THE CHEMISTRY OF LIFE. As you practice the exercises on each handout, you will be

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able to: CHEMISTRY OF LIFE: Develop and use a model to illustrate the hierarchical organization of

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Periodic Table Packet
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#1 ANSWERS

Directions: Answer the questions with the proper information using your notes, book, and the periodic table.

1. Define a family.

GROUPS ON THE TABLE ARE SOMETIMES CALLED FAMILIES. 2.

What is a period? A PERIOD IS A ROW ON THE PERIODIC TABLE 3.

What is the symbol for the following elements.

a. Magnesium Mg b.

Potassium K

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Periodic Table Packet #1

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Section 1.4 – Problem Solving in Chemistry.

The three steps to

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solving a numeric word problem are analyze, calculate, and evaluate. Analyze - identify the known and unknown. Calculate - solve the problem. Evaluate - is your answer reasonable? Always remember to put a unit!!

Chapter 1: Introduction to Chemistry

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Unit 13: Nuclear
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Key Ideas. Stability of
isotopes is based in the
ratio of neutrons and
protons in its nucleus.
Although most nuclei

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are stable, some are unstable and spontaneously decay, emitting radiation.
(3.1o)

Unit 13: Nuclear Chemistry Class Packet

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Semester 1 - AP Chemistry

CHEMISTRY MOLES

PACKET PAGE 3

CHEMISTRY

WORKSHEET # 1

MOLAR MASS (GRAM

MOLECULAR/FORMULA

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WEIGHTS) We know that grams are actually a measure of the mass of matter and not the weight. Mass is the quantity of

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one of the options to
accompany you later
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4 answers Air is a mixture of gases (and not a compound), about 78% being nitrogen (an element) , about 21% oxygen (another element), and all other gases present in much lesser amounts.

Answers about Chemistry

NGSS Regents
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Chemistry Practice
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and NYS Regents
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